# Rockstar Sealing

Chemwatch: 5570-71

Version No: 2.1 Safety Data Sheet according to Work Health and Safety Regulations (Hazardous Chemicals) 2023 and ADG requirements Chemwatch Hazard Alert Code: 2

Issue Date: **12/06/2024** Print Date: **12/06/2024** S.GHS.AUS.EN.E

# SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier	
Product name	Natural Finish Stone Sealer
Chemical Name	Not Applicable
Synonyms	Not Available
Proper shipping name	FLAMMABLE LIQUID, N.O.S. (contains n-butyl acetate and naphtha petroleum, heavy, hydrotreated)
Chemical formula	Not Applicable
Other means of identification	Not Available

# Relevant identified uses of the substance or mixture and uses advised against

Delevent identified vess	Protective treatment for mineral surfaces
Relevant identified uses	Use according to manufacturer's directions.

## Details of the manufacturer or supplier of the safety data sheet

Registered company name	Rockstar Sealing
Address	3 Spray Avenue Mordialloc VIC 3195 Australia
Telephone	+61 130 088 4418
Fax	+61 395 805 530
Website	www.rockstarsealing.com.au
Email	admin@rockstarsealing.com.au

#### Emergency telephone number

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Association / Organisation	Rockstar Sealing
Emergency telephone numbers	1300 88 44 18
Other emergency telephone numbers	Not Available

#### **SECTION 2 Hazards identification**

#### Classification of the substance or mixture

Poisons Schedule	S5
Classification <sup>[1]</sup>	Flammable Liquids Category 3, Aspiration Hazard Category 1, Serious Eye Damage/Eye Irritation Category 2B, Specific Target Organ Toxicity - Single Exposure (Narcotic Effects) Category 3
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI

Label elements

Hazard pictogram(s)		
Signal word	Danger	
Hazard statement(s)		
H226	Flammable liquid and vapour.	
H304	May be fatal if swallowed and enters airways.	
H320	Causes eye irritation.	
H336	May cause drowsiness or dizziness.	
AUH066	Repeated exposure may cause skin dryness and cracking.	
Precautionary statement(s) Pre	evention	
P210	Keen owny from heat het outgood, and the and other ignition outgood. No amplying	

# P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking. P271 Use only outdoors or in a well-ventilated area. P240 Ground and bond container and receiving equipment.

P241	Use explosion-proof electrical/ventilating/lighting/intrinsically safe equipment.	
Precautionary statement(s) Re	sponse	
P301+P310	IF SWALLOWED: Immediately call a POISON CENTER/doctor/physician/first aider.	
P331	Do NOT induce vomiting.	
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam to extinguish.	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
Precautionary statement(s) Sto	prage	
P403+P235	Store in a well-ventilated place. Keep cool.	
P405	Store locked up.	
Precautionary statement(s) Dis	ទposal	
P501	Dispose of contents/container to authorised bazardous or special waste collection point in accordance with any local regulation	

## **SECTION 3 Composition / information on ingredients**

#### Substances

See section below for composition of Mixtures

#### Mixtures

CAS No	%[weight]	Name
64742-48-9.	>70	naphtha petroleum, heavy, hydrotreated
123-86-4	5-25	n-butyl acetate
Not Available	5-15	Ingredients determined not to be hazardous
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L * EU IOELVs available	

# **SECTION 4 First aid measures**

#### Description of first aid measures

Eye Contact	<ul> <li>If this product comes in contact with the eyes:</li> <li>Wash out immediately with fresh running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>		
Skin Contact	<ul> <li>If skin contact occurs:</li> <li>Immediately remove all contaminated clothing, including footwear.</li> <li>Flush skin and hair with running water (and soap if available).</li> <li>Seek medical attention in event of irritation.</li> </ul>		
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> </ul>		
Ingestion	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> <li>Avoid giving milk or oils.</li> <li>Avoid giving alcohol.</li> <li>If spontaneous vomiting appears imminent or occurs, hold patient's head down, lower than their hips to help avoid possible aspiration of vomitus.</li> </ul>		

#### Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

## **SECTION 5 Firefighting measures**

## Extinguishing media

- Alcohol stable foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.

#### Special hazards arising from the substrate or mixture

Fire Incompatibility

Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Advice fo	r firefighters
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Advice for menginers	
Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear breathing apparatus plus protective gloves.</li> <li>Prevent, by any means available, spillage from entering drains or water course.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Liquid and vapour are flammable.</li> <li>Moderate fire hazard when exposed to heat or flame.</li> <li>Vapour forms an explosive mixture with air.</li> <li>Moderate explosion hazard when exposed to heat or flame.</li> <li>Combustion products include:</li> <li>carbon monoxide (CO)</li> <li>carbon dioxide (CO2)</li> <li>other pyrolysis products typical of burning organic material.</li> </ul>
HAZCHEM	•3Y

# **SECTION 6 Accidental release measures**

Personal precautions, protective equ	ipment and emergency procedures
See section 8	

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## Environmental precautions

See section 12

# Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Remove all ignition sources.</li> <li>Clean up all spills immediately.</li> <li>Avoid breathing vapours and contact with skin and eyes.</li> <li>Control personal contact with the substance, by using protective equipment.</li> </ul>
Major Spills	<ul> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>May be violently or explosively reactive.</li> <li>Wear breathing apparatus plus protective gloves.</li> </ul>

Personal Protective Equipment advice is contained in Section 8 of the SDS.

# **SECTION 7 Handling and storage**

Precautions for safe handling	
Safe handling	<ul> <li>DO NOT allow clothing wet with material to stay in contact with skin</li> <li>Avoid all personal contact, including inhalation.</li> <li>Wear protective clothing when risk of overexposure occurs.</li> <li>Use in a well-ventilated area.</li> <li>Prevent concentration in hollows and sumps.</li> </ul>
Other information	<ul> <li>Store in original containers in approved flammable liquid storage area.</li> <li>Store away from incompatible materials in a cool, dry, well-ventilated area.</li> <li>DO NOT store in pits, depressions, basements or areas where vapours may be trapped.</li> <li>No smoking, naked lights, heat or ignition sources.</li> </ul>

# Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>Metal drums/cans</li> <li>Packing as supplied by manufacturer.</li> <li>Plastic containers may only be used if approved for flammable liquid.</li> <li>Check that containers are clearly labelled and free from leaks.</li> <li>For low viscosity materials (i) : Drums and jerry cans must be of the non-removable head type. (ii) : Where a can is to be used as an inner package, the can must have a screwed enclosure.</li> <li>For materials with a viscosity of at least 2680 cSt. (23 deg. C)</li> <li>For manufactured product baying a viscosity of at least 250 cSt.</li> </ul>
Storage incompatibility	<ul> <li>n-Butyl acetate:</li> <li>reacts with water on standing to form acetic acid and n-butyl alcohol</li> <li>reacts with water on standing to form acetic acid and n-butyl alcohol</li> <li>reacts violently with strong oxidisers and potassium tert-butoxide</li> <li>is incompatible with caustics, strong acids and nitrates</li> <li>dissolves rubber, many plastics, resins and some coatings</li> <li>Low molecular weight alkanes are a type of chemical compounds that can be found in gases or liquids. These alkanes:</li> <li>Can cause a dangerous reaction with strong oxidizers, chlorine, chlorine dioxide, and dioxygenyl tetrafluoroborate when there is oxygen and heat present.</li> <li>Are incompatible with halogens.</li> <li>Can create static charges due to their low conductivity, leading to an accumulation of static charge.</li> <li>Esters react with acids to liberate heat along with alcohols and acids.</li> <li>Strong oxidising acids may cause a vigorous reaction with esters that is sufficiently exothermic to ignite the reaction products.</li> <li>Heat is also generated by the interaction of esters with caustic solutions.</li> <li>Flammable hydrogen is generated by mixing esters with alkali metals and hydrides.</li> <li>Avoid strong acids, bases.</li> </ul>

## SECTION 8 Exposure controls / personal protection

#### **Control parameters**

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Mater	ial name	TWA		STE	L	Peak	Notes
Australia Exposure Standards	naphtha petroleum, heavy, hydrotreated	Oil mi miner	st, refined al	5 mg/	m3	Not /	Available	Not Available	Not Available
Australia Exposure Standards	n-butyl acetate	n-Buty	/l acetate	150 p mg/m	pm / 713 3	950 ppm	mg/m3 / 200	Not Available	Not Available
Emergency Limits									
Ingredient	TEEL-1		TEEL-2				TEEL-3		
naphtha petroleum, heavy, hydrotreated	350 mg/m3 1,800 mg/m3					40,000 mg/m3			
n-butyl acetate	Not Available	Not Available					Not Available		
la sue di su é					Device d IDI				
Ingredient	Original IDLH				Revised IDL	н			
naphtha petroleum, heavy, hydrotreated	2,500 mg/m3			Not Available					
n-butyl acetate	1,700 ppm				Not Available				

#### Exposure controls

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk. Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.
Individual protection measures, such as personal protective equipment	
Eye and face protection	<ul> <li>Safety glasses with side shields.</li> <li>Chemical goggles. [AS/NZS 1337.1, EN166 or national equivalent]</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task.</li> </ul>
Skin protection	See Hand protection below
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gumboots, e.g. Rubber</li> <li>For esters:</li> <li>Do NOT use natural rubber, butyl rubber, EPDM or polystyrene-containing materials.</li> <li>The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.</li> <li>The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.</li> <li>Personal hygiene is a key element of effective hand care.</li> </ul>
Body protection	See Other protection below
Other protection	<ul> <li>Overalls.</li> <li>PVC Apron.</li> <li>PVC protective suit may be required if exposure severe.</li> <li>Eyewash unit.</li> <li>Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.</li> <li>For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets).</li> <li>Non sparking safety or conductive footwear should be considered. Conductive footwear describes a boot or shoe with a sole made from a conductive compound chemically bound to the bottom components, for permanent control to electrically ground the foot an shall dissipate static electricity from the body to reduce the possibility of ignition of volatile compounds.</li> </ul>

# Recommended material(s)

#### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

Natural Finish Stone Sealer

Material	СРІ
PE/EVAL/PE	A
PVA	A
TEFLON	A
BUTYL	С
BUTYL/NEOPRENE	С
HYPALON	С
NATURAL RUBBER	С
NEOPRENE	С
NEOPRENE/NATURAL	С
NITRILE	С
NITRILE+PVC	С

# **Respiratory protection**

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	A-AUS	-	A-PAPR-AUS / Class 1
up to 50 x ES	-	A-AUS / Class 1	-
up to 100 x ES	-	A-2	A-PAPR-2 ^

#### ^ - Full-face

 $\begin{array}{l} \mathsf{A}(\mathsf{All classes}) = \mathsf{Organic vapours}, \mathsf{B} \: \mathsf{AUS or} \: \mathsf{B1} = \mathsf{Acid gasses}, \mathsf{B2} = \mathsf{Acid gas or} \\ \mathsf{hydrogen cyanide(HCN)}, \: \mathsf{B3} = \mathsf{Acid gas or} \: \mathsf{hydrogen cyanide(HCN)}, \: \mathsf{E} = \mathsf{Sulfur} \\ \mathsf{dioxide(SO2)}, \: \mathsf{G} = \mathsf{Agricultural chemicals}, \: \mathsf{K} = \mathsf{Ammonia(NH3)}, \: \mathsf{Hg} = \mathsf{Mercury}, \: \mathsf{NO} = \\ \mathsf{Oxides of nitrogen}, \: \mathsf{MB} = \mathsf{Methyl bromide}, \: \mathsf{AX} = \mathsf{Low boiling point organic} \\ \mathsf{compounds}(\mathsf{below} \: \mathsf{65 degC}) \end{array}$ 

 Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.

PE	С	
PVC	С	
VITON/BUTYL	С	1

\* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final

selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors

such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

#### **SECTION 9** Physical and chemical properties

#### Information on basic physical and chemical properties

The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.

Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

Appearance	Slightly turbid liquid with petroleum odour; does not mix with water.		
Physical state	Liquid	Relative density (Water = 1)	0.81
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	126-200	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	24	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Flammable.	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

## **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

#### **SECTION 11 Toxicological information**

Information on toxicological effects

The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by sleepiness, reduced alertness, loss of reflexes, lack of co-ordination, and vertigo. The main effects of simple esters are irritation, stupor and insensibility. Headache, drowsiness, dizziness, coma and behavioural changes may occur. Inhalation hazard is increased at higher temperatures. Inhaling high concentrations of mixed hydrocarbons can cause narcosis, with nausea, vomiting and lightheadedness. Low molecular weight (C2-C12) hydrocarbons can irritate mucous membranes and cause incoordination, giddiness, nausea, vertigo, confusion, headache, appetite loss, drowsiness, tremors and stupor. Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.
Swallowing of the liquid may cause aspiration into the lungs with the risk of chemical pneumonitis; serious consequences may result. (ICSC13733) Accidental ingestion of the material may be damaging to the health of the individual.
Repeated exposure may cause skin cracking, flaking or drying following normal handling and use. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. The material may accentuate any pre-existing dermatitis condition 511r21?r

	There is some evidence to suggest that the material	may cause moderate inflammation	of the skin either following direct contact or after a
Eye	Instillation of isoparaffins into rabbit eyes produces of There is some evidence that material may produce of instillation. Severe inflammation may be expected w	only slight irritation. eye irritation in some persons and pr ith redness.	oduce eye damage 24 hours or more after
Chronic	Long-term exposure to respiratory irritants may resu Prolonged or repeated skin contact may cause dryin Substance accumulation, in the human body, may o exposure. There has been some concern that this material can Constant or exposure over long periods to mixed hy loss and anaemia, and reduced liver and kidney fun Chronic solvent inhalation exposures may result in r	It in airways disease, involving diffic ig with cracking, irritation and possib ccur and may cause some concern f in cause cancer or mutations but then drocarbons may produce stupor with ction. Skin exposure may result in dr pervous system impairment and liver	ulty breathing and related whole-body problems. le dermatitis following. ollowing repeated or long-term occupational e is not enough data to make an assessment. dizziness, weakness and visual disturbance, weight ying and cracking and redness of the skin. and blood changes. [PATTYS]
	τοχιριτή	IRRITATION	
Natural Finish Stone Sealer	Not Available	Not Available	
	τοχιείτα		
naphtha petroleum, heavy, bydrotreated		Eye: no advers	
nyaroneateu	Inhalation (Rat) LC50: >4.42 mg/L4h <sup>11</sup>	Skin: adverse e	ffect observed (irritating) <sup>L1</sup>
	Oral (Rat) LD50: >4500 mg/kg <sup>11</sup>		
	ΤΟΧΙΟΙΤΥ	IRRITATION	
	Dermal (rabbit) LD50: 3200 mg/kg <sup>[2]</sup>	Eye ( human): 3	300 mg * [PPG]
	Inhalation (Rat) LC50: 0.74 mg/l4h <sup>[2]</sup>	Eye (rabbit): 20	mg (open)-SEVERE
n-butyl acetate	Oral (Rabbit) LD50; 3200 mg/kg <sup>[2]</sup>	Eye (rabbit): 20	mg/24h - moderate
		Eye: no advers	e effect observed (not irritating) <sup>[1]</sup>
		Skin (rabbit): 50	00 mg/24h-moderate
		Skin: no advers	e effect observed (not irritating) <sup>[1]</sup>
Logond.	specified data extracted from RTECS - Register of 1	Foxic Effect of chemical Substances	
	Animal studies indicate that normal, branched and c		
NAPHTHA PETROLEUM, HEAVY, HYDROTREATED	paraffins is inversely proportional to the carbon chait to be present in mineral oil, n-paraffins may be abso The major classes of hydrocarbons are well absorbe hydrocarbons are ingested in association with fats ir the gut lymph, but most hydrocarbons partly separal Petroleum contains aromatic (benzene, toluene, eth many detrimental health effects, including, cancer, tt Animal testing shows breathing in petroleum causes humans. Similarly, exposure to gasoline over a lifetir Most studies involving gasoline have shown that gas subjects (such as in petrol service station attendants Animal studies show concentrations of toluene (>0.1 toxicity to the nervous system of the foetus. Other st Prolonged contact with petroleum may result in skin materials.	yclic paraffins are absorbed from the In length, with little absorption above ribed to a greater extent than iso- or ad into the gastrointestinal tract in va in the diet. Some hydrocarbons may a te from fats and undergo metabolism yl benzene, napthalene) and aliphati umour formation, hearing loss, and r is tumours of the liver and kidney; the me can cause kidney cancer in anime soline does not cause genetic mutatif (%) can cause developmental effects trudies show no adverse effects on the inflammation and make the skin mo	e gastrointestinal tract and that the absorption of n- C30. With respect to the carbon chain lengths likely cyclo-paraffins. rious species. In many cases, the hydrophobic appear unchanged as in the lipoprotein particles in i in the gut cell. c hydrocarbons (n-hexane), which can result in revous system toxicity. se are however not considered to be relevant in als, but the relevance in humans is questionable. on, including all recent studies in living human s such as lower birth weight and developmental e foetus. re sensitive to irritation and penetration by other
NAPHTHA PETROLEUM, HEAVY, HYDROTREATED	<ul> <li>Animal studies indices indices and control and the carbon chait to be present in mineral oil, n-paraffins may be abso. The major classes of hydrocarbons are well absorbe hydrocarbons are ingested in association with fats in the gut lymph, but most hydrocarbons partly separat Petroleum contains aromatic (benzene, toluene, eth many detrimental health effects, including, cancer, tt Animal testing shows breathing in petroleum causes humans. Similarly, exposure to gasoline over a lifetim Most studies involving gasoline have shown that gas subjects (such as in petrol service station attendants Animal studies show concentrations of toluene (&gt;0.1 toxicity to the nervous system of the foetus. Other st Prolonged contact with petroleum may result in skin materials.</li> <li>Generally,linear and branched-chain alkyl esters are and most tissues throughout the body. Following hyd Oral acute toxicity studies have been performed in vitro us saturated carboxylic acids: methyl acetate, butyl acet these substances are not genotoxic.</li> <li>The JEFCA Committee concluded that the substance seters of aliphatic acyclic primary alcohols and aliph average maximum levels of 200 mg/kg. Higher level hard candy.</li> <li>The material may produce severe irritation to the eye produce conjunctivitis.</li> </ul>	yclic paraffins are absorbed from the In length, with little absorption above rived to a greater extent than iso- or an the diet. Some hydrocarbons may a the from fats and undergo metabolism yl benzene, napthalene) and aliphati umour formation, hearing loss, and r a tumours of the liver and kidney; the me can cause kidney cancer in anim soline does not cause genetic mutati s). (%) can cause developmental effect: tudies show no adverse effects on the inflammation and make the skin mo drolysis the component alcohols and 1 of the 67 esters of aliphatic acyclic his group of esters is demonstrated sing the following esters of aliphatic state, butyl stearate and the structure es in this group would not present si atic linear saturated carboxylic acides s of use (up to 3000 mg/kg) are perr e causing pronounced inflammation. d or repeated exposure and may pro- skin.	e gastrointestinal tract and that the absorption of n- C30. With respect to the carbon chain lengths likely cyclo-paraffins. rious species. In many cases, the hydrophobic appear unchanged as in the lipoprotein particles in i in the gut cell. c hydrocarbons (n-hexane), which can result in ervous system toxicity. se are however not considered to be relevant in als, but the relevance in humans is questionable. on, including all recent studies in living human s such as lower birth weight and developmental e foetus. re sensitive to irritation and penetration by other hols and carboxylic acids in the intestinal tract, blood carboxylic acids are metabolized primary alcohols and aliphatic linear ally related isoamyl formate and demonstrates that afety concerns at the current levels of intake the are generally used as flavouring substances up to nitted in food categories such as chewing gum and Repeated or prolonged exposure to irritants may boduce on contact skin redness, swelling, the
NAPHTHA PETROLEUM, HEAVY, HYDROTREATED	<ul> <li>Animal studies indicate that nonlinal, branched and c</li> <li>paraffins is inversely proportional to the carbon chait</li> <li>to be present in mineral oil, n-paraffins may be abso</li> <li>The major classes of hydrocarbons are well absorbe</li> <li>hydrocarbons are ingested in association with fats if</li> <li>the gut lymph, but most hydrocarbons partly separat</li> <li>Petroleum contains aromatic (benzene, toluene, eth</li> <li>many detrimental health effects, including, cancer, tt</li> <li>Animal testing shows breathing in petroleum causes</li> <li>humans. Similarly, exposure to gasoline over a lifetin</li> <li>Most studies involving gasoline have shown that gas</li> <li>subjects (such as in petrol service station attendants</li> <li>Animal studies show concentrations of toluene (&gt;0.1</li> <li>toxicity to the nervous system of the foetus. Other st</li> <li>Prolonged contact with petroleum may result in skin</li> <li>materials.</li> <li>Generally,linear and branched-chain alkyl esters are</li> <li>and most tissues throughout the body. Following hyd</li> <li>Oral acute toxicity studies have been performed in vitro us</li> <li>saturated carboxylic acids: methyl acetate, butyl ace</li> <li>these substances are not genotoxic.</li> <li>The JEFCA Committee concluded that the substance</li> <li>esters of aliphatic acyclic primary alcohols and aliph</li> <li>average maximum levels of 200 mg/kg. Higher level</li> <li>hard candy.</li> <li>The material may produce severe irritation to the eye</li> <li>produce conjunctivitis.</li> <li>The material may cause skin irritation after prolonge</li> <li>production of vesicles, scaling and thickening of the</li> </ul>	yclic paraffins are absorbed from the In length, with little absorption above rived to a greater extent than iso- or ad into the gastrointestinal tract in van In the diet. Some hydrocarbons may is the from fats and undergo metabolism yl benzene, napthalene) and aliphati umour formation, hearing loss, and r a tumours of the liver and kidney; the me can cause kidney cancer in anim soline does not cause genetic mutati is). 1%) can cause developmental effects udies show no adverse effects on th inflammation and make the skin mo e hydrolysed to their component alco drolysis the component alcohols and I of the 67 esters of aliphatic acyclic his group of esters is demonstrated sing the following esters of aliphatic tate, butyl stearate and the structurate es in this group would not present si: atic linear saturated carboxylic acids s of use (up to 3000 mg/kg) are perr e causing pronounced inflammation. ad or repeated exposure and may pro- skin. <b>Carcinogenicity</b>	e gastrointestinal tract and that the absorption of n- C30. With respect to the carbon chain lengths likely cyclo-paraffins. rious species. In many cases, the hydrophobic appear unchanged as in the lipoprotein particles in in the gut cell. c hydrocarbons (n-hexane), which can result in ervous system toxicity. se are however not considered to be relevant in als, but the relevance in humans is questionable. on, including all recent studies in living human s such as lower birth weight and developmental e foetus. re sensitive to irritation and penetration by other hols and carboxylic acids in the intestinal tract, blood carboxylic acids are metabolized primary alcohols and aliphatic linear saturated by oral LD50 values greater than 1850 mg/kg bw acyclic primary alcohols and aliphatic linear ally related isoamyl formate and demonstrates that afety concerns at the current levels of intake the are generally used as flavouring substances up to nitted in food categories such as chewing gum and Repeated or prolonged exposure to irritants may bduce on contact skin redness, swelling, the
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Legend: X – Data either not available or does not fill the criteria for classification - Data available to make classification

Toxicity

Natural Finish Stone Sealer	Endpoint	Test Duration (hr)	Species	Value	Source
	Not Available	Not Available	Not Available	Not Available	Not Available
	Endpoint	Test Duration (hr)	Species	Value	Source
naphtha petroleum, heavy,	EC50(ECx)	48h	Crustacea	>0.002mg/l	2
hydrotreated	EC50	96h	Algae or other aquatic plants	64mg/l	2
	EC50	48h	Crustacea	>0.002mg/l	2
	Endpoint	Test Duration (hr)	Species	Value	Source
	LC50	96h	Fish	17- 19mg/L	4
n-butyl acetate	EC50	72h	Algae or other aquatic plants	246mg/l	2
	EC50	48h	Crustacea	32mg/l	1
	EC50(ECx)	96h	Fish	18mg/l	2
Legend:	Extracted from	1. IUCLID Toxicity Data 2. Europe E	ECHA Registered Substances - Ecotoxicological Inform	nation - Aquatic Toxicity	4. US EF

Harmful to aquatic organisms, may cause long-term adverse effects in the aquatic environment. DO NOT discharge into sewer or waterways.

## Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
n-butyl acetate	LOW	LOW
Bioaccumulative potential		
Ingredient	Bioaccumulation	
n-butyl acetate	LOW (BCF = 14)	
Mobility in soil		
Ingredient	Mobility	
n-butyl acetate	LOW (Log KOC = 20.86)	

# **SECTION 13 Disposal considerations**

Waste treatment methods	
Product / Packaging disposal	<ul> <li>DO NOT allow wash water from cleaning or process equipment to enter drains.</li> <li>It may be necessary to collect all wash water for treatment before disposal.</li> <li>In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.</li> <li>Where in doubt contact the responsible authority.</li> <li>Recycle wherever possible.</li> <li>Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.</li> <li>Dispose of by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).</li> <li>Decontaminate empty containers.</li> </ul>

# **SECTION 14 Transport information**

# Labels Required

Marine Pollutant	NO
HAZCHEM	•3Y
Land transport (ADG)	

Land transport (7.20)				
14.1. UN number or ID number	1993	1993		
14.2. UN proper shipping name	FLAMMABLE LIQUID,	N.O.S. (contains n-butyl acetate and naphtha petroleum, heavy, hydrotreated)		
14.3. Transport hazard class(es)	Class Subsidiary Hazard	3 Not Applicable		
14.4. Packing group	Ш			
14.5. Environmental hazard	Not Applicable			

Special provisions	223 274
Limited quantity	5 L
	Special provisions Limited quantity

## Air transport (ICAO-IATA / DGR)

		7					
14.1.	UN number	1993	1993				
14.2.	UN proper shipping name	Flammable liquid, n.o.s. * (contains	ammable liquid, n.o.s. * (contains n-butyl acetate and naphtha petroleum, heavy, hydrotreated)				
		ICAO/IATA Class 3					
14.3.	Transport nazard	ICAO / IATA Subsidiary Hazard Not Applicable					
	01033(03)	ERG Code	3L				
14.4.	Packing group	Ш					
14.5.	Environmental hazard	Not Applicable					
		Special provisions		A3			
		Cargo Only Packing Instructions		366			
		Cargo Only Maximum Qty / Pack		220 L			
14.6. Special precautions for user	Passenger and Cargo Packing In	structions	355				
	Passenger and Cargo Maximum	Qty / Pack	60 L				
	Passenger and Cargo Limited Qu	antity Packing Instructions	Y344				
	Passenger and Cargo Limited Ma	aximum Qty / Pack	10 L				

#### Sea transport (IMDG-Code / GGVSee)

14.1. UN number	1993				
14.2. UN proper shipping name	FLAMMABLE LIQUID,	FLAMMABLE LIQUID, N.O.S. (contains n-butyl acetate and naphtha petroleum, heavy, hydrotreated)			
14.3. Transport hazard class(es)	IMDG Class IMDG Subsidiary Ha	3 azard Not Applicable			
14.4. Packing group	III				
14.5 Environmental hazard	Not Applicable				
14.6. Special precautions for user	EMS Number Special provisions Limited Quantities	F-E , S-E 223 274 955 5 L			

## 14.7.1. Transport in bulk according to Annex II of MARPOL and the IBC code Not Applicable

## 14.7.2. Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
naphtha petroleum, heavy, hydrotreated	Not Available
n-butyl acetate	Not Available

#### 14.7.3. Transport in bulk in accordance with the IGC Code

Product name	Ship Type
naphtha petroleum, heavy, hydrotreated	Not Available
n-butyl acetate	Not Available

## **SECTION 15 Regulatory information**

#### Safety, health and environmental regulations / legislation specific for the substance or mixture

## naphtha petroleum, heavy, hydrotreated is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals

Australian Inventory of Industrial Chemicals (AIIC)

Chemical Footprint Project - Chemicals of High Concern List

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs - Not Classified as Carcinogenic

#### n-butyl acetate is found on the following regulatory lists

Australia Hazardous Chemical Information System (HCIS) - Hazardous Chemicals

Australian Inventory of Industrial Chemicals (AIIC)

# Additional Regulatory Information

Not Applicable

#### National Inventory Status

National Inventory	Status
Australia - AIIC / Australia Non- Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	No (naphtha petroleum, heavy, hydrotreated; n-butyl acetate)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

#### **SECTION 16 Other information**

Revision Date	12/06/2024
Initial Date	19/01/2023

#### **SDS Version Summary**

Version	Date of Update	Sections Updated
3.1	20/02/2023	Toxicological information - Acute Health (eye), Toxicological information - Acute Health (skin), Toxicological information - Acute Health (swallowed), Physical and chemical properties - Appearance, Toxicological information - Chronic Health, Hazards identification - Classification, Disposal considerations - Disposal, Exposure controls / personal protection - Engineering Control, Ecological Information - Environmental, Firefighting measures - First Fighter (fire/explosion hazard), First Aid measures - First Aid (inhaled), First Aid measures - First Aid (swallowed), Handling and storage - Handling Procedure, Composition / information on ingredients - Ingredients, Exposure controls / personal protection - Personal Protection (other), Exposure controls / personal protection - Personal Protection (ther), Exposure controls / personal protection - Personal Protection (eye), Exposure controls / personal protection - Personal Protection (hands/feet), Accidental release measures - Spills (major), Handling and storage - Storage (storage incompatibility), Identification of the substance / mixture and of the company / undertaking - Use
3.1	24/02/2023	Toxicological information - Acute Health (eye), Composition / information on ingredients - Ingredients, Handling and storage - Storage (suitable container)

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### Definitions and abbreviations

- PC TWA: Permissible Concentration-Time Weighted Average
- PC STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit.
- IDLH: Immediately Dangerous to Life or Health Concentrations
- ES: Exposure Standard
- OSF: Odour Safety Factor
- NOAEL: No Observed Adverse Effect Level
- LOAEL: Lowest Observed Adverse Effect Level
   TLV: Threshold Limit Value
- LOD: Limit Of Detection
- OTV: Odour Threshold Value
- BCF: BioConcentration Factors
- BEI: Biological Exposure Index
- DNEL: Derived No-Effect Level
- PNEC: Predicted no-effect concentration
- AIIC: Australian Inventory of Industrial Chemicals
- DSL: Domestic Substances List
- NDSL: Non-Domestic Substances List
   IECSC: Inventory of Existing Chemical Substance in China
- EINECS: European INventory of Existing Commercial chemical Substances
- ELINCS: European List of Notified Chemical Substances
- NLP: No-Longer Polymers
- ENCS: Existing and New Chemical Substances Inventory
- KECI: Korea Existing Chemicals Inventory
- NZIOC: New Zealand Inventory of Chemicals

- PICCS: Philippine Inventory of Chemicals and Chemical Substances
   TSCA: Toxic Substances Control Act
- TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas
- NCI: National Chemical Inventory
   FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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